TABLE OF CONTENT

INTRODUCTION 5
  Scope 5

General Design Consideration 6

Liquid-liquid Coalescer 8

Principles of a liquid-liquid Coalescer 10
  1. Droplet Capture 10
  2. Droplet Coalescence 12
  3. Stokes Settling With Coalesced Droplets 15

Type of Coalescer Media 21

Plate Coalescer 21

Matrix Coalescer 24

Polypropylene Packs 26

Mesh Coalescer 29
Coalescer Cartridges

Liquid-Gas Coalescer

Ratings/Sizing

DEFINITIONS

NOMENCLATURE

THEORY OF THE DESIGN

Liquid-Liquid Coalescer Design

A. Stokes Settling

B. Direct Interception

Liquid-Gas Coalescer Design

A. Gravity Settling

B. Media Velocity

C. Annular Velocity

APPLICATION

Example 1: Liquid-Gas Horizontal Coalescer Design

Example 2: Liquid-Gas Vertical Coalescer Design

Example 3: Liquid-Liquid Coalescer Design

Example 4: Liquid-Liquid Coalescer Design

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REFERENCES 80

LIST OF TABLE

Table 1. Coalescing media 14
Table 2. Coalescing media and their applications 14
Table 3. Comparison Between Separation time and Droplet Size 16
Table 4. performance of several media type coalescer 27
Table 5. Types of Liquid/Gas Separators 36
Table 6. Typical sources and characteristics that can generate dispersion 52
Table 7. Type of coalesce with their coefficient 56

LIST OF FIGURE

Figure 1: Vertical coalesce 7
Figure 2: Horizontal coalesce 8
Figure 3: Three steps coalescing 10
Figure 4: Zones where different coalescing mechanisms apply 11
Figure 5: Coalescing in the medium 15
Figure 6: Horizontal liquid–liquid coalescer configuration 19
Figure 7: Vertical liquid–liquid coalescer configuration 20
Figure 8: Plate Coalescer pack 22
Figure 9 Several Type of Plate Coalescer 23
Figure 10: Matrix Coalescer Packs 24

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INTRODUCTION

Scope

This guideline covers the basic elements of coalescer design in sufficient detail to allow a practicing engineer to design a coalescer with the suitable size including diameter, length media velocity, and terminal settling velocity.

For coalescers, as with any process equipment, successful sizing and selection is always a combination of empirical observation/experience and analytical modeling. Of the three steps in coalescing – droplet capture, combining of the collected droplets, and gravity separation of the enlarged droplets – the first and the last can be modeled with good accuracy and repeatability. The modeling of the middle and the actual coalescing step is a complex function of surface tension and viscous effects, droplet momentum, and the dynamics of the sizes of the droplets in the dispersion.

The design of coalescer may be influenced by factors, including process requirements, economics and safety. In this guideline there are tables that assist in making these factored calculations from the various reference sources. Include in this guideline is a calculation spreadsheet for the engineering design. All the important parameters used in this guideline are explained in the definition section which helps the reader understand the meaning of the parameters and / or the terms used.

The theory section explains source, type of coalesces and its characteristic of droplet, treated and untreated coalescer and how to calculate sizing and selection of the coalescer. The application of the coalescer theory with an example will help the practicing engineer understand the coalescer and be ready to perform the actual design of the coalescer.
General Design Consideration

The biggest development in recent years is the widespread recognition that the actual performance of a separator may fall far short of the theoretical performance due to the actual flow patterns within the vessel being far from the ideal. It has, however, been helped by two visualization techniques computational fluid dynamics (CFD) and physical modeling, which vividly show what can go wrong and how to correct it.

The following factors must be determined before beginning separator design.

1. Gas and liquids flow rates (minimum, average, and peak).
2. Operating and design pressures and temperatures.
3. Surging or slugging tendencies of the feed streams.
4. Physical properties of the fluids, such as density, viscosity, and compressibility.
5. Designed degree of separation

The most important areas to ensure a separator performs to design are as follow.

1. Correct inlet nozzle sizing and a good inlet device (momentum breaker).
2. Primary fluid distribution–distribution plates to translate the reduced but still high velocities from the inlet device into quiescent flows in a liquid–liquid separator body, or distribution plates either side of a vane pack (downstream is best as upstream ones shatter droplets unnecessarily) or other gas demister.
3. Intermediate fluid distribution when necessary.

Coalescer is a mechanical process vessel with wet-able, high-surface area packing on which liquid droplets consolidate for gravity separation from a second phase (for example gas or immiscible liquid), where small particles of one liquid phase must be separated or removed from a large quantity of another liquid phase. The coalescers might be designed vertically or horizontally.

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The vertical design is used to separate water from hydrocarbons when the interfacial tension is greater than 3 dyne/cm. The separation stage is achieved using hydrophobic separator cartridges that provide an effective barrier to aqueous coalesced drops, but allow hydrocarbon to pass through them. The separator cartridges can be stacked below the coalescers for the most efficient utilization of the separator medium. This configuration only applies to the separation of water or aqueous contaminants from hydrocarbons.

After leaving the coalescing stage, the large aqueous coalesced drops and hydrocarbon then flow axially in a downward direction and the flow direction is from the outside of the separator to the inside. The large coalesced drops are repelled by the separators and are collected in the bottom sump. The purified hydrocarbon passes through the separators and exits at the top of the housing. The aqueous phase in the collection sump can be drained manually on a periodic basis or equipped with an automatic level control and drain system.

Figure 1: Vertical coalescer
In the horizontal configuration, a settling zone achieves separation by gravity. This configuration is used when the interfacial tension is less than 3 dyne/cm or for the separation of oil from the water phase. The coalescer housing contains a settling zone that relies on the difference in densities between the coalesced droplets and the bulk fluid.

This configuration can be used for both hydrocarbon from water and water from hydrocarbon separation, but the location of the collection sump and outlet nozzle will need to be reversed. For the case of removal of hydrocarbon from water, a collection sump is located at the top of the housing and the purified water leaves at the bottom outlet nozzle. The sump can be drained manually on a periodic basis or equipped with an automatic level control and drain system.

Figure 2. Horizontal coalescer
Liquid-liquid coalescer

Efficient liquid-liquid separation are an integral part of many industrial processes. Whether engineering a new coalescer vessel, or debottlenecking an existing separator, full knowledge and understanding of the basic principles involved are required. Often overlooked are the capabilities of properly selected and designed internals for the enhancement of simple gravity separation. This guideline describes the use of various media and methods employed for decades to increase plant productivity. Typical applications include:

- Removal of Bottlenecks in existing Decanters and Three Phase Separators.
- Reduction in New Vessel Sizes – Up to five times relative to gravity settling alone.
- Improvements in Product Purity – Carry-over entrainment reduced to 1 ppm and less.
- Compliance with Environmental Regulations – Cost effective solutions to wastewater treatment and oil spill cleanups.

When two liquids are immiscible, or non-soluble in one another, they can form either an emulsion or a colloidal suspension. In either of these mixtures, the dispersed liquid forms droplets in the continuous phase. In a suspension, the droplets are less than one micron in diameter and the liquids cannot readily be separated with the technologies described here. Fortunately, in the chemical and hydrocarbon process industries droplet sizes are typically greater than this and/or the purities required can be achieved without addressing the ultra-light colloidal component of the stream.

Liquid–liquid separations may require the use of special equipment when the drop sizes are small, typically in the range of 1 to 50 μm in size. These fluid systems are classified as stable emulsions, and often conventional bulk separators with mist pads or plate-type internals will not be effective. High-efficiency liquid–liquid coalescers have been developed to break these emulsions and provide improved separation.

Coalescers are typically manufactured as either pads or cartridge filters that have been designed especially to take small droplets in an emulsion and grow them into large drops that are separated more easily. This process is accelerated over natural coalescing by the fibers present in coalesce media that force the contact of small...
droplets, thereby promoting the coalescing process. The pore gradient of coalescer medium is constructed so that the inlet medium has fine pore sizes that increase in size with the flow direction.

Liquid–liquid coalescer are constructed from polymer and fluoro polymer materials that have been optimized to separate the most difficult emulsions with interfacial tensions as low as 0.5 dyne/cm. This coalescer can be used with a broad range of applications. It can process aggressive chemicals and handle demanding operating conditions while providing the highest level of performance.

Basis for Sizing and Selection

A preliminary procedure for determining how difficult it is to separate two immiscible liquids involves the performance of a simple field test. A representative sample of the emulsion is taken from a process pipeline or vessel. It is either put it in a graduated cylinder in the lab or, if it is under pressure, in a clear flow-through sample tube with isolation valves. The time required to observe a clean break between phases is noted. If the continuous phase has a viscosity less than 3 centipoise, then Stokes Law says the following:

<table>
<thead>
<tr>
<th>Separation Time</th>
<th>Emulsion Stability</th>
<th>Droplet Size, Microns</th>
</tr>
</thead>
<tbody>
<tr>
<td>&lt; 1 Minutes</td>
<td>Very Week</td>
<td>&gt;500</td>
</tr>
<tr>
<td>&lt; 10 Minutes</td>
<td>Week</td>
<td>100-500</td>
</tr>
<tr>
<td>Hours</td>
<td>Moderate</td>
<td>40-100</td>
</tr>
<tr>
<td>Days</td>
<td>Strong</td>
<td>1-40</td>
</tr>
<tr>
<td>Weeks</td>
<td>Very Strong</td>
<td>&lt;1 (Colloidal)</td>
</tr>
</tbody>
</table>

Fortunately, the experienced designer with knowledge of the application, equipment, and physical properties can often estimate the strength of the emulsion and determine which medium will be successful. A more definitive approach, and one that is often needed to provide a process warranty, is the use of an on-site pilot unit.
Liquid-liquid coalescer performance is often rated in parts per million of dispersed phase allowable in the continuous phase effluent. Even trace amounts of contaminants such as emulsifiers and chemical stabilizers can have dramatic effects on results at these levels. In a pilot program, several alternate media are provided to the customer so that their performance can be documented on the actual process stream, thereby taking into account the effects of any particulates or surfactants present.

For liquid-liquid coalescers, as with any process equipment, successful sizing and selection is always a combination of empirical observation/experience and analytical modeling. Of the three steps in coalescing – droplet capture, combining of the collected droplets, and gravity separation of the enlarged droplets – the first and the last can be modeled with good accuracy and repeatability. The modeling of the middle and the actual coalescing step is a complex function of surface tension and viscous effects, droplet momentum, and the dynamics of the sizes of the droplets in the dispersion. This has been done successfully in porous media.

The successful design of a liquid-liquid coalescer starts with knowledge of the source of the emulsion and the stream’s physical properties.

1. Typically the minimum droplet size is estimated to be between 75 to 300μm.
2. Large vessels and long residence times are required to settle fine droplets and ensure laminar flow.
3. When the average droplet is greater than roughly 1/2 millimeter (500 microns), an open gravity settler is appropriate.
4. Successful gravity separation downstream of a coalesce element depends primarily on vessel geometry.
5. A boot is desirable when the amount of dispersed phase is <15% v/v where the control of the interface level is linear with the volume of dispersed phase discharged.
6. A dispersed phase velocity of 10 inches (254 mm)/minute is desirable to allow disengagement of the continuous phase, while keeping the boot diameter <40% of the diameter of the horizontal portion to minimize the necessity for weld pads.

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7. The most common applications for coalescers in vertical flow are extraction/liquid-liquid absorption towers and entrainment knockout installations where the available plot plans in the plant are at a premium.

8. The coalescer is located downstream of the interface so that entrained continuous phase is removed from the dispersed.

9. Some recommendations that the liquid loading in a vertical wash tower be limited to at most 1.6 ft/min of the dispersed phase.

10. With the installation of a coalescer this can safely be increased to 2 ft/min (15 gpm/ft²) thereby decreasing the cross-sectional area of the column by 20 to 40%.

11. It is important economically to keep the L/D ratio in the range of 3 to 5.

12. It is typical and desirable that coalesced droplets emerge from media that operates either on Intra- Media Stokes Settling or Direct Interception at a size of from 500 to 1,000 microns.

13. The vessel length necessary for inlet distribution devices upstream of the media (such as sparger pipes, ‘picket fences’, and perforated plates used to assure uniform flow through the media and the depth of the typical coalescer element itself with supports is typically 1 to 1.5 D.

There are some benefits from modern coalescer technology:

1. Reduce size and cost of new liquid-liquid separators
2. Improve product purity in existing installations
3. Debottleneck existing reflux drums
4. Reduce downstream corrosion caused by corrosive liquid carryover
5. Reduce losses of glycol, amine and other valuable chemicals
6. Improve fractionation tower operation and reduce maintenance
7. Reduce fugitive VOC emissions

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There are some limitations to consider in liquid-liquid coalesce:

1. Generally, the solids range that liquid–liquid coalescers can operate economically with disposable filters is <10 ppm. Above this level of solids, further pretreatment will be required.

2. If the clarified stream leaving the coalescer is then cooled, condensation of a previously dissolved contaminant can occur, leading to a hazy fluid at the lower temperature.

3. Coalescers typically will have a service life of 1 to 2 years when protected adequately by prefiltration.

4. For liquid–liquid coalescers constructed from glass fiber medium, the problem of surfactant disarming must also be considered and for low IFT emulsion systems (<20 dyne/cm) they will not operate efficiently for separation.

High-efficiency liquid–liquid coalescers are finding increasing applications in industry where problematic emulsions exist. They are used to protect equipment, to recover valuable streams, and to meet environmental discharge limits. Some examples are given.

1. Pipeline Condensate - to separate the difficult emulsions to purify both water and hydrocarbon phases.
2. Produced Water - offers a reliable way to meet environmental limits. Petrochemical Final Products. Can lead to water condensation and off specification hazy final products.
3. Caustic Treating, to be an effective solution to recover the caustic carryover
Operating principles of a liquid-liquid coalescer

Liquid-liquid coalescers are used to accelerate the merging of many droplets to form a lesser number of droplets, but with a greater diameter. This increases the buoyant forces in the Stokes Law equation. Settling of the larger droplets downstream of the coalesce element then requires considerably less residence time. Coalescers exhibit a three-step method of operation as depicted in Figure 3.

1. **Droplet Capture**

The first step of coalescing is to collect entrained droplets primarily either by Intra-Media Stokes Settling or Direct Interception. Figure 4 gives the useful zones of separation for various mechanisms. Solids can increase the stability of an emulsion and removing solids can make coalescing easier. Generally, this step can be achieved by a separate cartridge filter system or by a regenerable backwash filter system for high levels of solids. In addition, the filtration stage protects the coalesce and increases service life. This step also initiates the coalescence of the hydrocarbon droplets, thereby enhancing the separation capabilities of the system.
A general rule with Direct Interception is that the size of the target should be close to the average sized droplet in the dispersion. Finer coalescing media allow for the separation of finer or more stable emulsions (Table 1). Note that fine media will also capture or filter fine solid particulates from the process stream. Therefore, unless the emulsion is very clean, an upstream duplex strainer or filter is needed to protect a high efficiency coalescer.

Droplet capture, the first step in liquid-liquid coalescing, is the most important. The next two sections describe the formulas used for the collection mechanisms of Intra-Media Stokes Settling and Direct Interception.

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Figure 4: Zones where different coalescing mechanisms apply
2. Droplet Coalescence

The second step is to combine, aggregate, or coalesce captured droplets. Increasing the tendency for droplets to adhere to a medium, increases the probability that subsequent droplets will have the opportunity to strike and coalesce with those that already have been retained.

Whether a coalescer medium is hydrophilic (likes water) or oleophilic (likes oil) depends on the solid/liquid interfacial tension between it and the dispersed phase. In general an organic dispersed phase ‘wets’ organic (that is plastic or polymeric) media, as there is a relatively strong attraction between the two, while an aqueous dispersed phase preferably ‘wets’ inorganic media, such as metals or glass. This aids in the coalescence step as the droplets adhere to the media longer. Also assisting coalescing is the density of media: lower porosities yield more sites available for coalescing. In the case of yarns and wools, capillary forces are also important for retaining droplets.

Once several droplets are collected on a plate, wire, or fiber, they will tend to combine in order to minimize their interfacial energy. Predicting how rapidly this will occur without pilot testing is very difficult to do. Judgments of the proper volume, and therefore residence time, in the coalescers are guided by experience and the following properties:

Coalescing Media:
1. Media/Dispersed Phase Interfacial Tension
2. Porosity
3. Capillarity

Liquid Phases:
1. Continuous/Dispersed Interfacial Tension
2. Continuous/Dispersed Density Difference
3. Continuous Phase Viscosity
4. Superficial Velocity
Different coalescing materials have been found suitable for different applications. Commercially, fiberglass seems to be the most popular medium due to its availability and low cost. Table 1 below shows different coalescing medias with their different surface properties, cost and fouling properties. In addition to the numerous coalescing media, Table 2 presents some coalescing media and their industrial applications with regard to the nature of emulsions they separate (emulsion source), flowrate and maximum droplet diameter.

Coalescers work better in laminar flow for several reasons.

1. Droplets will stay in the streamlines around a wire or fiber target.
2. High fluid velocities overcome surface tension forces and strip droplets out of the coalescer medium. This results in reentrainment in co-current flow and prevents droplets from rising/sinking in counter-current flow.
3. Slower velocities result in greater residence time in the media and therefore more time for droplet-to-target impact, droplet-to-droplet collisions, and Intra-Media Stokes Settling.

The coalescence mechanism can be described by the following steps.

1. Droplet adsorption to fiber
2. Translation of droplets to fiber intersections by bulk flow
3. Coalescence of two droplets to form one larger droplet
4. Repeated coalescence of small droplets into larger droplets at fiber intersections
5. Release of droplets from fiber intersections due to increased drag on adsorbed droplets caused by bulk flow
6. Repeat of steps 1–5 with progressively larger droplet sizes and more open media porosity
Table 1. Coalescing media

<table>
<thead>
<tr>
<th>Media</th>
<th>Surface Characteristic</th>
<th>Porosity</th>
<th>Size</th>
<th>Fouling/Cost</th>
<th>Spacing/Crimps</th>
</tr>
</thead>
<tbody>
<tr>
<td>Metal/plastic corrugated sheets</td>
<td>hydophilicity, oleophilicity</td>
<td>98-99%</td>
<td>9,5-25,4 mm</td>
<td>Low/Low</td>
<td></td>
</tr>
<tr>
<td>Wire/plastic mesh</td>
<td>hydophilicity, oleophilicity</td>
<td>95-99%</td>
<td>0,05-0,279 mm</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Wire wool</td>
<td>Hydophilicity</td>
<td>95-99%</td>
<td>0,05-0,279 mm</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Wire/polymer co-knits</td>
<td>Oleophilicity</td>
<td>94-98%</td>
<td>21-35 μm</td>
<td>High/High</td>
<td></td>
</tr>
<tr>
<td>Wire/fiberglass co-knits, glass mat</td>
<td>Hydophilicity</td>
<td>92-96%</td>
<td>9-10 μm</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

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Table 2. Coalescing media and their applications

<table>
<thead>
<tr>
<th>Media</th>
<th>Source</th>
<th>Max. Droplet Diameter (μm)</th>
<th>Flow Range (gpm/ft²)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Corrugated sheets</td>
<td>Separators with coarse emulsions and static mixers.</td>
<td>40-1000</td>
<td>15-75 (35-180 m³/hr/m²)</td>
</tr>
<tr>
<td>Wire mesh, wire wool</td>
<td>Extraction columns, Distillation tower feeds, impeller mixers.</td>
<td>20-300</td>
<td>7,5-45 (35-180 m³/hr/m²)</td>
</tr>
<tr>
<td>Co-knits of wire and polymer</td>
<td>Steam stripper bottoms, Caustic wash drums, High pressure drop mixing valves.</td>
<td>10-200</td>
<td>7,5-45 (35-180 m³/hr/m²)</td>
</tr>
<tr>
<td>Glass mat, co-knits of wire and fiberglass</td>
<td>Haze from cooling in bulk liquid phase, Surfactants giving, Emulsions with very low interfacial tension.</td>
<td>1-25</td>
<td>7,5-45 (35-180 m³/hr/m²)</td>
</tr>
</tbody>
</table>
DEFINITIONS

**Air standard** - Air having a temperature of (20°C), a relative humidity of 36 percent, and under a pressure of 14.70 PSIA. The gas industry usually considers (16°C) as the temperature of standard air.

**Annular velocity** - the actual flow rate divided by the annulus area. Modeled as a linear function with vertical distance, and the annular velocity is zero at the bottom of the cartridge and increases to a maximum value at the top of the cartridge.

**Annulus** - A ring-like part or, the orifice of a hollow die, through which extruded metal flows from the press.

**Coalescence** - Liquid particles in suspension that unite to create particles of a greater volume.

**Coalescer** - a mechanical process vessel with wettable, high-surface area packing on which liquid droplets consolidate for gravity separation from a second phase (for example gas or immiscible liquid), where small particles of one liquid phase must be separated or removed from a large quantity of another liquid phase.

**Conventional Gas-Liquid Separator** - In this Standard, the term "Conventional Gas-Liquid Separator" is referred to vertical or horizontal separators in which gas and liquid are separated by means of gravity settling with or without a mist eliminating device.

**Demister Mist Extractor** - A device installed in the top of scrubbers, separators, tray or packed vessels, etc. to remove liquid droplets entrained in a flowing gas stream.

**Entrainment** - A process in which the liquid boils so violently that suspended droplets of liquid are carried in the escaping vapor.

**Extraction column** - Vertical-process vessel in which a desired product is separated from a liquid by countercurrent contact with a solvent in which the desired product is preferentially soluble.

**Filter** - A piece of unit operation equipment by which filtration is performed.
Gas filter - A device used to remove liquid or solid particles from a flowing gas stream.

Immiscible - Not capable of mixing (as oil and water).

Knock-Out - A separator used for a bulk separation of gas and liquid.

Laminar flow - Streamlined flow of a fluid where viscous forces are more significant than inertial forces, generally below a Reynolds number of 2000.

Liquid-liquid extraction - The removal of a soluble component from a liquid mixture by contact with a second liquid, immiscible with the carrier liquid in which the component is preferentially soluble.

Media velocity - the actual flow rate divided by the coalescer filter area.

Mesh - The "mesh count" (usually called "mesh"), is effectively the number of openings of a woven wire filter per 25 mm, measured linearly from the center of one wire to another 25 mm from it.

mmscfd - Abbreviation for million standard cubic feet per day; usually refers to gas flow.

Mud sump - Upstream area in a process vessel where, because of a velocity drop, entrained solids drop out and are collected in a sump.

Process analyzer - An instrument for determining the chemical composition of the substances involved in a chemical process directly, or for measuring the physical parameters indicative of composition.

Separator - a cylindrical or spherical vessel used to isolate the components in mixed streams of fluids.

Stokes' law - the law that the force that retards a sphere moving through a viscous fluid is directly proportional to the velocity of the sphere, the radius of the sphere, and the viscosity of the fluid.
Suspension - a system consisting of a suspension of solid particles in a liquid

Terminal Velocity - The velocity at which a particle or droplet will fall under the action of gravity, when drag force just balances gravitational force and the particle (or droplet) continues to fall at constant velocity.

NOMENCLATURES

|Δρ| Absolute value of the difference between the densities of the continuous and dispersed phase, lb/ft$^3$
A Cross sectional area, ft$^2$
A$_{ann}$ Cross sectional annular area, ft$^2$
A$_{med}$ Media area for one coalesce, ft$^2$
C Drag coefficient.
C$_1$ Sheet coefficient
d Droplet Diameter, in
d Droplet Diameter, microns
Dc Diameter of house, ft
d$_p$ Droplet diameter, microns or ft
E Effective Length Multiplier
g Gravitational constant, ft/s$^2$
h Corrugated plate spacing or structured packing crimp height, in
K Kuwabara's Hydrodynamic Factor
L Element length required for removal of all droplets, in
M Mass flow at standard condition, lb/s
N Number of coalescers
Q$_a$ Actual system flow rate, ft/s
Ql Liquid/liquid emulsion flow, US GPM
Q$_s$ Standard system flow rate, ft/s
R$_c$ Radius of coalescer end cap, ft
R$_h$ Radius of the housing, ft
S$_g$ Specific gravity
t$_{dr}$ Droplet Rise Time, s
t$_r$ Droplet Residence Time, s
V$_C$ Coalescer volume, ft$^3$
v$_{max}$ Emulsion velocity, ft/s
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